# Fast and sensitive ball SAW hydrogen sensor with porous Pd-Pt alloy film

多孔性 Pd-Pt 合金膜を用いた高速かつ高感度な ボール SAW 水素センサ

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## 1. Introduction

To use  $H_2$  gas as common energy source, fast and sensitive sensor is required for early detection of the leakage. For monitoring use, sensors working at room temperature (RT) were studied and developed, based on resistive and surface acoustic wave (SAW) sensors[1-4]. In this situation, a ball SAW sensor, that enhances the sensitivity by multiple-roundtrip propagation of collimated SAW beam, realized the widest detection range (10 ppm to 100%) [5] with Pd-Ni alloy film [6]. However, their response time at low concentration were not still sufficiently short, typically ~100 s to 100 ppm H<sub>2</sub>.

To shorten the response time, both the dissociation and diffusion of hydrogen atoms into the sensitive film should be promoted. They may be enhanced by increasing surface area such as porous structure and by alloying catalytic element such as Pt [7]. In addition, Pt alloying Pd was proof against the phase transition [7]. Thus, although porous Pd-Pt alloy film is promising for H<sub>2</sub> gas sensing, it has not been applied to the ball SAW sensor. In this study, we develop the ball SAW sensor with porous Pd-Pt alloy film for realizing fast and sensitive hydrogen sensor working at RT.

## 2. Measurement implementation

Typical sputterd-film structure with respect to a substrate temperature and an Ar pressure was reported in reference 8. Based on it, we deposited porous Pd-Pt film at RT and 2.7 Pa using an rf magnetron sputtering. Pt-alloying was performed by co-sputtering of Pd (50W) and Pt (20W). As a result, Pd-20%Pt alloy film was fabricated on a langasite ball SAW device ( $\phi$ 3.3 mm, 150 MHz) with a thickness of 40 nm.

A schematic diagram of an apparatus is shown in Fig. 1. In order to adjust  $H_2$  concentration,  $H_2$ - $N_2$  mixed standard gas was diluted with  $N_2$  gas (G1 grade), using the digital mass flow controllers (MFCs) so as to be total flow rate of 50 ccm. The

sensor was set in a sensor cell with a dead volume of  $0.005 \text{ cm}^3$  and the temperature kept constant at  $35^{\circ}$ C. An amplitude at the  $10^{\text{th}}$  turn were measured using a digital quadrature detector (DQD) [9].



Fig. 1 Schematics of measurement apparatus

## 3. Result

A result of  $H_2$  measurement is shown in Fig. 2. Clear amplitude response was observed from 1000 ppm to 20 ppm. S/N of the response to 20 ppm  $H_2$  was 7.6, comparable to the best result of the ball SAW sensor with Pd-30%Ni alloy film [5]. Moreover, the response time from 10% to 90% at 1000 ppm was 4.1 s as shown in Fig. 3.



Fig. 2 Measurement of  $H_2$  (a) amplitude change at the 10th turn. (b)  $H_2$  concentration in  $N_2$ .



Fig. 3 Response to 1000 ppm H<sub>2</sub>.

Fig. 4 shows the concentration dependence of an amplitude change, plotted with log-log scales. The amplitude response is expressed by the function of the square root of the concentration. The reason may be explained by Sievelts's law, where the solubility of a diatomic gas in metal is proportional to the square root of the partial pressure [7]. Although similar dependence were reported in sensitive resistive sensor [1], it was for the first time in SAW sensor. The detection limit at S/N of 3 was 3.7 ppm.



Fig. 4 Evaluation of detection limit. Symbols represent experimental results. Solid and dotted lines represent the fitting of square root function and triple value of rms noise, respectively.

#### 4. Discussion

We compare the concentration dependence of the response time with those of pioneering studies, as shown in Fig. 5. Solid circle ( $\bigcirc$ ) represents the result of this study. Upper triangle ( $\triangle$ )[2], square ( $\square$ )[3], and lower triangle ( $\bigtriangledown$ )[4] represent the results of planar SAW sensors. Cross ( $\times$ )[1] represents that of resistive sensor. The response time of the ball SAW sensor was 5 times shorter than the best value of planar SAW sensor and resistive sensor. This result may be explained by improving a permeation of H<sub>2</sub> gas into the film due to porous structure, shorting an equilibrium time due to tens nm thickness, and promoting dissociation of H<sub>2</sub> molecule due to Pt alloying.



Fig. 5 Comparison of  $H_2$  concentration dependence of 10%-90% response time with pioneering studies. • represents this study.  $\triangle$ ,  $\Box$ , and  $\bigtriangledown$  represent planar SAW sensors with Pd film [2], Pd-Cu phthalocyanine composite film [3], and InOx film covered with thin Pt layer [4], respectively.  $\times$ represents resistive sensor with carbon nanotube with Pd nano-particles [1].

#### 5. Conclusion

It was shown that the fastest and most sensitive  $H_2$  sensor working at room temperature could be realized using the ball SAW sensor with porous Pd-Pt alloy film.

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